

An Easy Stereoselective Approach to Polyunsaturated Silylated Sulfides

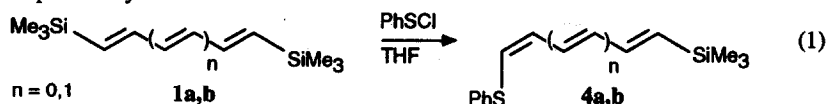
Vito Flandanese* and Luigia Mazzone

Centro CNR di Studio sulle Metodologie Innovative di Sintesi Organiche, Dipartimento di Chimica,
Università di Bari, via Amendola 173, 70126 Bari, Italy

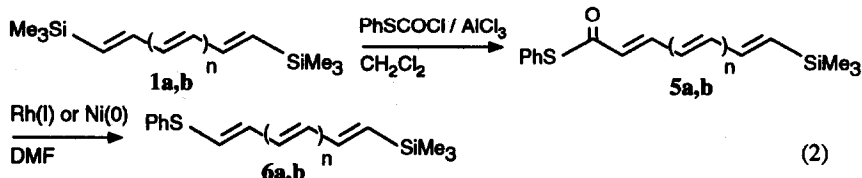
Abstract: Stereodefined monosilylated dienyl and trienylsulfides have been easily prepared by means of chemoselective substitution reactions on unsaturated disilyl derivatives.

We have recently pursued a program of research that utilizes simple and readily available organosilicon derivatives¹. In the frame of such a program herein we report a very convenient route to monosilylated sulfides with a conjugate diene and triene structure, a class of compounds having silicon and sulfur functionalities which are both useful^{2,3}. The procedure is based upon chemoselective substitution reactions on unsaturated disilyl derivatives 1.

Accordingly, we started our investigation performing the reactions of compounds 1 with the readily accessible phenylsulfenyl chloride and we found that the reaction of 1,4-disilylated diene 1a (n=0) or 1,6-disilylated triene 1b (n=1) with PhSCl in THF at room temperature (eq. 1) led to the monosubstitution product 4a with inversion of configuration (Z,E/E,E = 95/5, by capillary GLC analysis) (70% yield) or 4b (Z,E,E/E,E,E = 90/10) (51% yield)⁴, in accordance with the electrophilic substitution of the silyl group with bromine in simple alkenylsilanes⁵.



Moreover, to obtain the stereoisomers of E configuration, we adopted a variation of the strategy (eq. 2).



In particular, we performed the substitution reaction of compounds 1 with the commercially available S-phenyl carbonochloridothioate in the presence of AlCl₃, obtaining the corresponding monosilylated thioesters 5a (85% yield) or 5b (77% yield) with high retention of configuration (stereoisomeric purities \geq 98%). A straightforward decarbonylation reaction^{6,7}, performed with RhCl(PPh₃)₃⁶ on compound 5a or with Ni(PPh₃)₄⁷ on compound 5b, led to the corresponding sulfide 6a (E,E/Z,E = 96/4) (72% yield) or 6b (E,E,E/Z,E,E = 85/15) (65% yield). The stereochemical course of this reaction is in agreement with the fact

that the Rh(I)- and Ni(0)-promoted decarbonylation of α,β -unsaturated thioesters proceeds with retention of configuration^{6,7}. It is worth noting that the Ni(0)-decarbonylation of **5a** leads to lower yields of product **6a**, whereas a low value of stereoselectivity is observed in the Rh(I)-decarbonylation reaction of **5b**.

Representative procedures for the synthesis of **4a** and **6a**. A THF solution (10 mL) of PhSCl (3.0 mmol) was added at room temperature under nitrogen to a solution of **1a** (2.5 mmol) in THF (5 mL). After 2 h, the mixture was quenched with saturated aqueous NH_4Cl and the usual workup followed by flash chromatography over silica gel (petroleum ether as eluant) gave the product **4a**⁸ in 70% yield.

A CH_2Cl_2 solution (10 mL) of PhSCoCl (5 mmol) was added, under nitrogen, to a cold (0°C) stirred suspension of anhydrous AlCl_3 (5 mmol) in CH_2Cl_2 (5 mL). After 10 min at 0°C, the clear solution was transferred *via* syringe, under nitrogen, to the addition funnel of a three-necked flask containing a CH_2Cl_2 solution (10 mL) of **1a** (5 mmol). After complete addition at 0°C and stirring for 1 h, the mixture was quenched with saturated aqueous NH_4Cl and the residue purified by distillation to give 1.11 g (85% yield) of **5a** (b.p. 128-130°C, 1 mbar)⁸. Decarbonylation of **5a**: $\text{RhCl}(\text{PPh}_3)_3$ (1.67 mmol) and **5a** (1.67 mmol) were dissolved in DMF (20 mL) under nitrogen and the system was maintained at 80°C for 1 h. After cooling to room temperature, water was added, and the mixture was extracted with ether. The solvent was evaporated, the residue taken up in a minimum of petroleum ether and passed through a short Florisil column, in order to remove the complex. The product **6a**⁸ was isolated in 72% yield after flash chromatography.

In conclusion the process described here, because of its procedural simplicity, appears to be a useful route to conjugated monosilylated sulfides, which in principle could be easily converted into a variety of dienyl or trienylsilanes, by means of cross-coupling reactions performed in the presence of transition metal catalysts² and with the phenylthio group acting as a leaving group. Moreover, the synthesis of compounds **5** is also important in its own right since it should provide a convenient preparation of unsaturated thioesters.

Acknowledgements: This work was supported by National Research Council of Italy (C.N.R.), Progetto Finalizzato Chimica Fine II and by Ministero dell'Università e della Ricerca Scientifica e Tecnologica.

REFERENCES AND NOTES

- Babudri, F.; Fiandanese, V.; Marchese, G.; Naso, F. *J. Chem. Soc., Chem. Commun.* **1991**, 237-239.
- Babudri, F.; Fiandanese, V.; Naso, F. *J. Org. Chem.* **1991**, *56*, 6245-6248; Babudri, F.; Fiandanese, V.; Naso, F.; Punzi, A. *Synlett* **1992**, 221-223.
- Fiandanese, V.; Marchese, G.; Mascolo, G.; Naso, F.; Ronzini, L.; *Tetrahedron Lett.* **1988**, *29*, 3705-3708.
- Pegram, J.J.; Anderson, C.B. *Tetrahedron Lett.* **1988**, *29*, 6719-6720.
- The reaction of **1a** with PhSCl in benzene or in CH_2Cl_2 led to a low amount of the monosubstitution product, together with a low amount of the 1,4-disilylated unsaturated sulfide, derived from the addition of phenylsulphenyl chloride and subsequent elimination of HCl instead of Me_3SiCl .
- Miller, R.B.; McGarvey, G. *J. Org. Chem.* **1979**, *44*, 4623-4633.
- Osakada, K.; Yamamoto, T.; Yamamoto, A. *Tetrahedron Lett.* **1987**, *28*, 6321-6324.
- Wenkert, E.; Chianelli, D. *J. Chem. Soc., Chem. Commun.* **1991**, 627-628; Tiecco, M.; Testaferri, L.; Tingoli, M.; Chianelli, D.; Montanucci, M. *Synthesis*, **1984**, 736-738.
- 4a**: ¹H-NMR (CDCl_3 , 200 MHz) δ : 0.24 (s, 9H), 6.09 (d, $J=18.2$ Hz, 1H), 6.32 (d, $J=9.3$ Hz, 1H), 6.4 (dd, $J=9.3, 9.3$ Hz, 1H), 7.07 (dd, $J=18.2, 9.3$ Hz, 1H), 7.29-7.48 (m, 5H_{arom}) ppm. ¹³C-NMR (CDCl_3 , 50.3 MHz) δ : -1.27, 125.24, 126.71, 129.13, 129.22, 129.27, 132.20, 136.61, 138.74 ppm. MS:m/e 234 (M^+ , 11), 219 (4), 157 (26), 109 (14), 73 (100). **5a**: ¹H-NMR (CDCl_3) δ : 0.09 (s, 9H), 6.23 (d, $J=15.2$ Hz, 1H), 6.5 (d, $J=18.1$ Hz, 1H), 6.65 (dd, $J=18.1, 9.2$ Hz, 1H), 7.23 (dd, $J=15.2, 9.2$ Hz, 1H), 7.35-7.60 (m, 5H_{arom}) ppm. ¹³C-NMR (CDCl_3) δ : -1.64, 127.36, 129.16, 129.23, 129.38, 134.59, 140.91, 143.12, 147.71, 188.28 ppm. MS:m/e 262 (M^+ , 2), 247 (2), 182 (13), 153 (100), 125 (33), 109 (44), 95 (27), 73 (83). **6a**: ¹H-NMR (CDCl_3) δ : 0.08 (s, 9H), 5.78 (d, $J=17.7$ Hz, 1H), 6.46 (d, $J=14.7$ Hz, 1H), 6.52-6.70 (m, 2H), 7.20-7.45 (m, 5H_{arom}) ppm. ¹³C-NMR (CDCl_3) δ : -1.33, 127.07, 127.67, 129.12, 129.21, 130.18, 133.10, 134.03, 142.20 ppm. MS:m/e 234 (M^+ , 9), 219 (5), 157 (23), 109 (16), 73 (100).